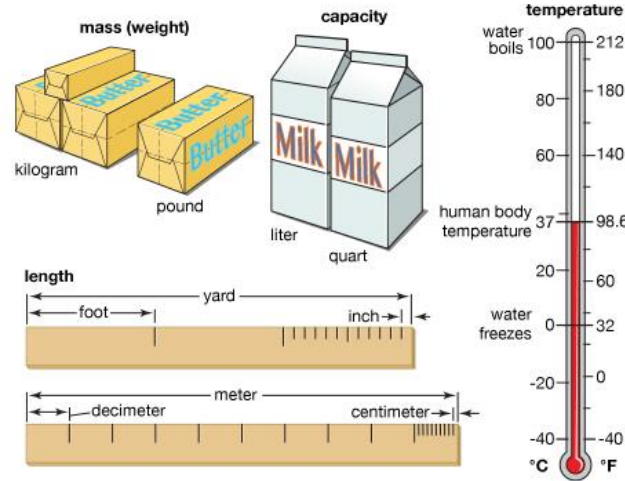


# The Metric System



Customary and international system (SI) units



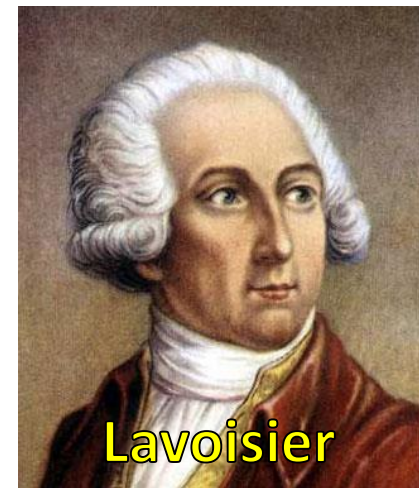
kilometer centimeter  
10<sup>+3</sup> micrometer  
femto nano centi  
mega atto 10<sup>-6</sup> nanometer  
10<sup>+6</sup> kilo meter



# The International System of Units

# Origin of the Metric System

- Idea of standardized system of measurement based on the **decimal** was first proposed as early as ~1670.
- The first practical implementation was carried out by French Revolutionaries towards the end of the 18th century.
- In 1790 a **committee** (including mathematicians **Laplace** and **Legendre**, and chemist **Lavoisier**) was appointed to **develop a unified, natural, universal system of measurement**.



It was called the "**metric**" system (French for *measure*).

# Metric System Basics

- The metric system was built around three base units that corresponded to a **certain kind of measurement**:
  - Length = **meter**
  - Volume = **liter**
  - Weight (Mass) = **gram**
- The **base units** were derived from the **natural world**: the *dimensions of the Earth* and *properties of water*.
- Decimal multiplicative prefixes were added to base units to make up the **full range** of metric system:
  - **Milli** + **meter** = millimeter
  - **Kilo** + **gram** = Kilogram
- Historically, prototypes (“originals”) of base units were kept in the *Archives Nationales in France* with copies manufactured and distributed among other 17 countries - members of The Metre Convention of 1875.

# Fundamental Units

By design, the **base units** used in the Metric System must be **realizable**, ideally with reference to natural phenomena.

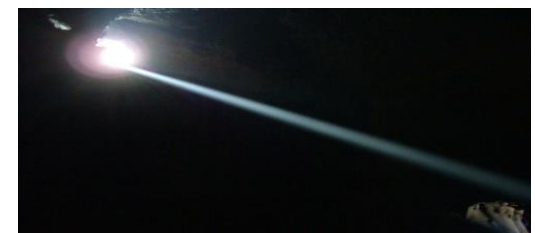
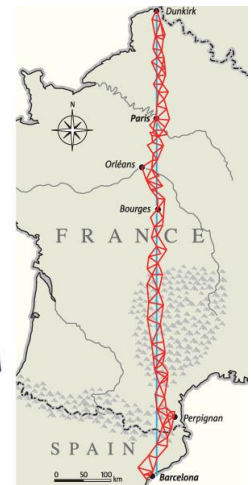
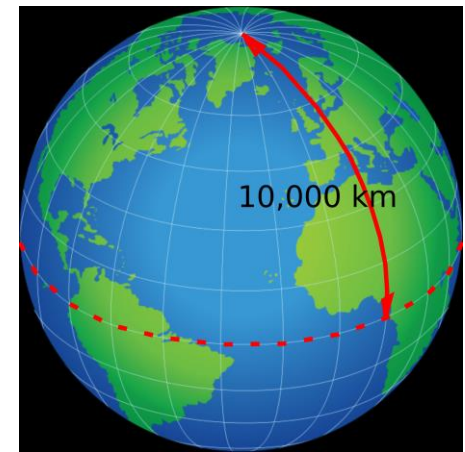
As Metric System evolved into the **SI system**, **seven mutually independent fundamental units** have been selected:

1. **Meter** (length)
2. **Kilogram** (mass)
3. **Second** (time)
4. **Kelvin** (temperature)
5. **Ampere** (electric current)
6. **Candela** (luminous intensity)
7. **Mole** (amount of elementary entities like atoms or molecules)



# The Meter, m (Metric, SI)

- **Original definition (1791): one ten millionth (1/10,000,000) of the quarter of the Earth's meridian** (distance between the **North Pole** and the **Equator** through Paris was determined based on a Pierre Méchain and Jean-Baptiste Delambre 1792-1798 survey of the length of the Earth's meridian between Dunkirk (51°N) and Barcelona (41°N) through Paris).
- **1799: platinum bar**, known as the **mètre des Archives**. The International Metre Commission in Paris (1870-72, 1875): new “**metric prototypes**” made of **90% platinum and 10% iridium**.
- **1983: One meter is the distance traveled by a ray of light through a vacuum in 1/299,792,458 second.** The definition in terms of the speed of light means that the meter can be realized using any light source of known frequency, rather than defining a "preferred" source.

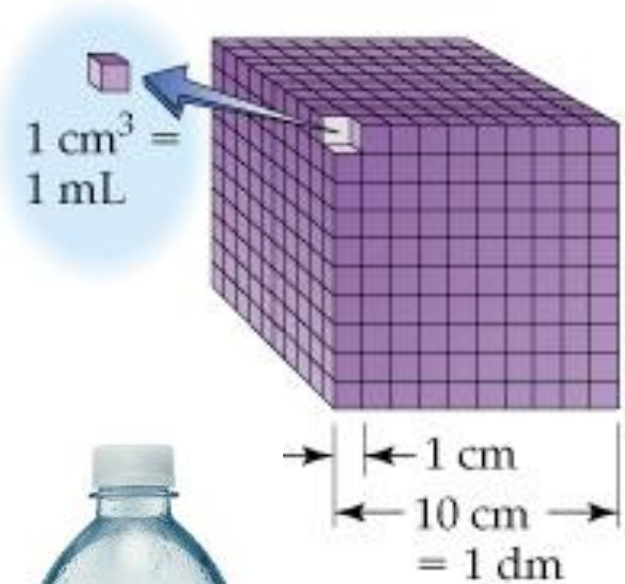


# The Liter, L (Metric, non-SI)

- **Unit of volume, one of three original base units** in metric system circa 1799.
- Volume = Length x Width x Height
- **The liter is equal to 1 cubic decimeter** (10×10×10 centimeters) or 1/1,000 cubic meter.
- One liter of liquid water has a mass of *almost exactly* one kilogram.
- Most commonly used for **fluids** and **solids that can be poured** (which are measured by the capacity or size of their container).

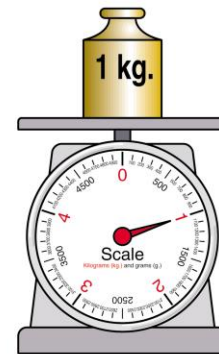
$$1\text{ L} = 1\text{ dm}^3 = 1000\text{ cm}^3$$

$$1\text{ cm}^3 = 1\text{ mL}$$



# The Gram and the Kilogram, g and kg (Metric, SI)

- **Mass:** the amount of matter in an object.
- **The gram, 1795:** the mass of one cubic centimeter of water at the melting point of water.
- The **original prototype kilogram** manufactured in 1799 had a mass equal to the mass of 1.000025 liters of water at 4°C.
- **1875:** **The International Prototype Kilogram,** a cylinder of platinum/platinum-iridium alloy.
- Kilogram is the only SI unit that is **still directly defined by an artifact.**
- **2005:** International Committee for Weights and Measures recommended that the kilogram **be redefined in terms of a fundamental constant of nature.**



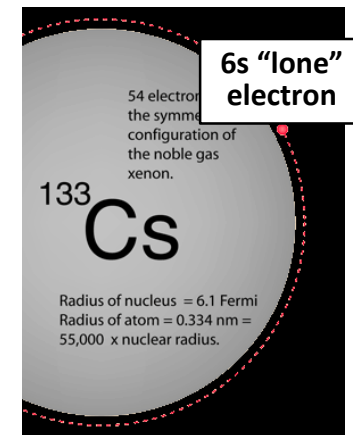
1 kilogram = 1000 grams



# The Second, s (Metric, SI)



- Earliest documented scientific use of **second as a unit of time**: ~1000 by Persian scholar al-Biruni.
- **Carl Friedrich Gauss, 1832**: proposed to use second as a **base unit**.
- **Original definition**: **1/86,400 of a mean solar day** (the Earth's rotation is slowing down, so in this definition a second was not a constant...)
- International Astronomical Union, 1952-1956 redefinition: the fraction 1/31,556,925.9747 of the tropical year 1900.
- **Current definition** (developed by General Conference on Weights and Measures in 1967 and refined in 1997): definition using atomic clocks as **the duration of 9,192,631,770 periods of the radiation corresponding to the transition between the two hyperfine levels of the ground state of the caesium-133 atom at rest at a temperature of 0 K.**
- SI prefixes starting from millisecond are commonly used to measure time less than a second (submultiples); non-SI units *minutes, hours, days, Julian years, Julian centuries, and Julian millennia* are used to measure multiples of second.

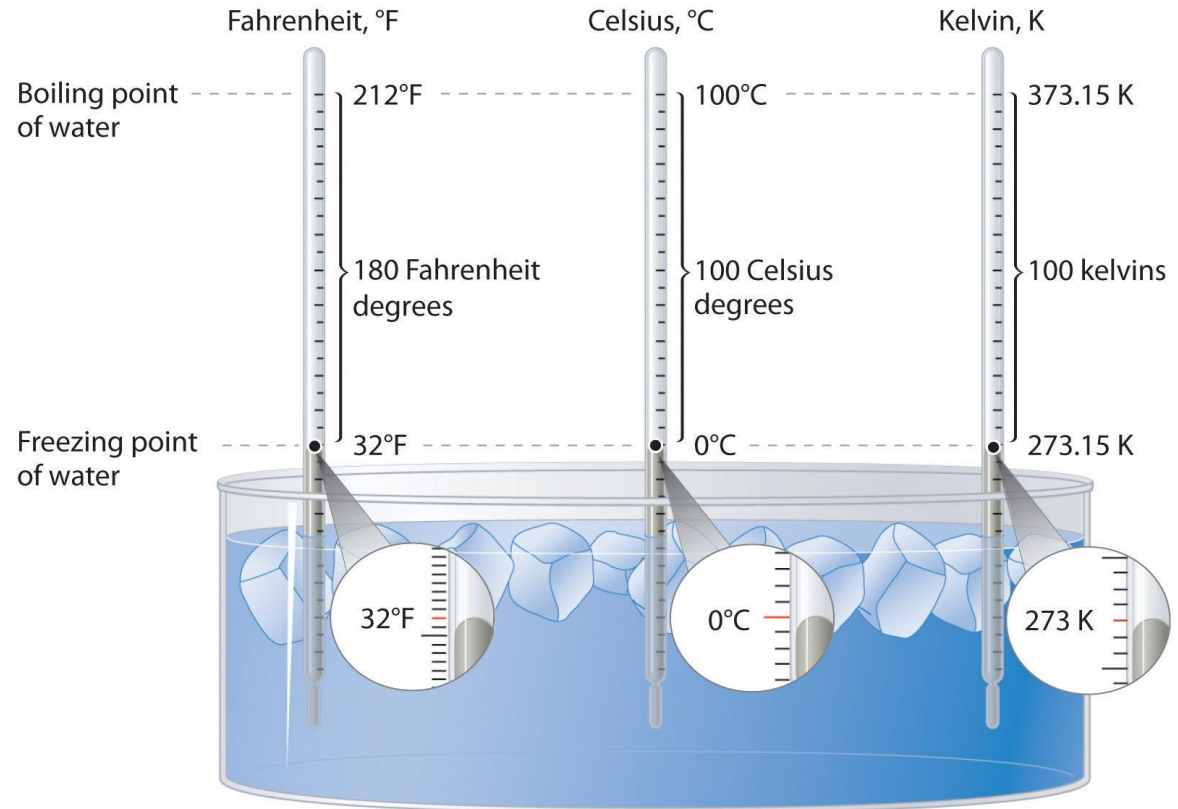




# The Kelvin, K (SI)

- **Unit of thermodynamic temperature** (*absolute temperature*).

- **Original (1743):** the **centigrade scale** (renamed “Celsius” in 1948) is obtained by assigning **0°C to the freezing point of water** and **100°C to the boiling point of water** both at a pressure of one standard atmosphere with *mercury* being the working material.



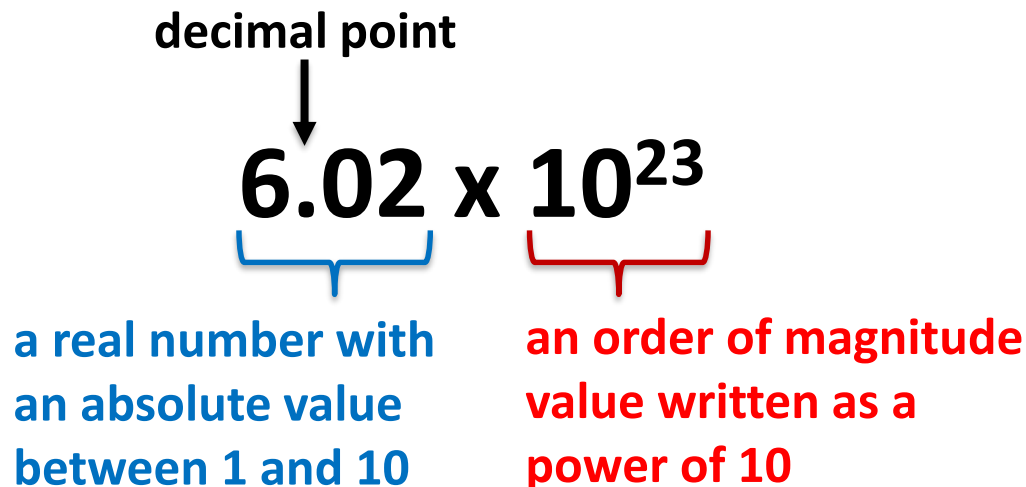
- **Current (1967):**  **$1/273.16$  of the thermodynamic temperature of the triple point of water** (triple point of water = 273.16 K = 0.01°C by definition).

# Prefixes in Metric System

Prefix	Symbol	Factor	
tera	T	1000000000000	$10^{12}$
giga	G	1000000000	$10^9$
mega	M	1000000	$10^6$
kilo	k	1000	$10^3$
hecto	h	100	$10^2$
deca	da	10	$10^1$
<b>(none)</b>	<b>(none)</b>	<b>1</b>	<b><math>10^0</math></b>
deci	d	0.1	$10^{-1}$
centi	c	0.01	$10^{-2}$
milli	m	0.001	$10^{-3}$
micro	μ	0.000001	$10^{-6}$
nano	n	0.000000001	$10^{-9}$
pico	p	0.000000000001	$10^{-12}$

# Scientific Notation

- **Scientific notation** (also referred to as "*standard form*" or "*standard index form*") is a way of writing numbers that are too big or too small to be conveniently written in decimal form.



- **When using SI system**, order of magnitude is restricted to follow standard SI prefixes; the resulting number representation is called "*engineering notation*".

# What is the order of the metric system?

- King Henry Died by Drinking Chocolate Milk

larger



smaller

- King: **Kilo**
- Henry: **Hecto**
- Died: **Deca**
- By: **Base** (m, L, g)
- Drinking: **Deci**
- Chocolate: **Centi**
- Milk: **Milli**



# Examples

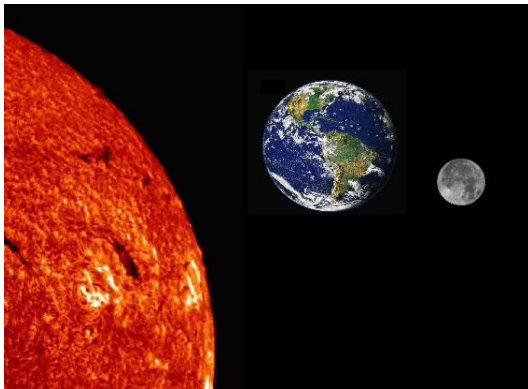
Any US paper currency note (\$1, \$5, \$10, \$20) has a mass of 1 g; the mass of a nickel is 5 g; the mass of a penny is 2.5 grams.



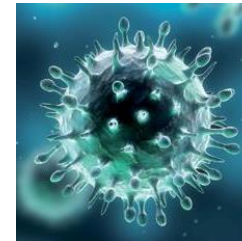
A typical doorknob is ~1 m high.



The mass of the Earth is  $6 \times 10^{24}$  kg; the mass of the Moon is  $7.3 \times 10^{22}$  kg; the mass of the Sun is  $1.99 \times 10^{30}$  kg.



Diameter of Influenza virus is ~20 nm.



Typical airport runway length is 3.35 km; Boeing 767 jet is 64 m long.



The diameter of a CD or a DVD is 12 cm; the diameter of the center hole is 15 mm.



# Imperial-Metric Equivalences

- Units of Weight

- 1 oz = 28.35 g
- 1 lb = 454 g
- 2.2 lb = 1 kg



- Units of Capacity

- 1.06 qt = 1 L
- 1 gal = 3.79 L



- Units of Length

- 1 in = 2.54 cm
- 3.28 ft = 1 m
- 1.09 yd = 1 m
- 1 mi = 1.61 km

