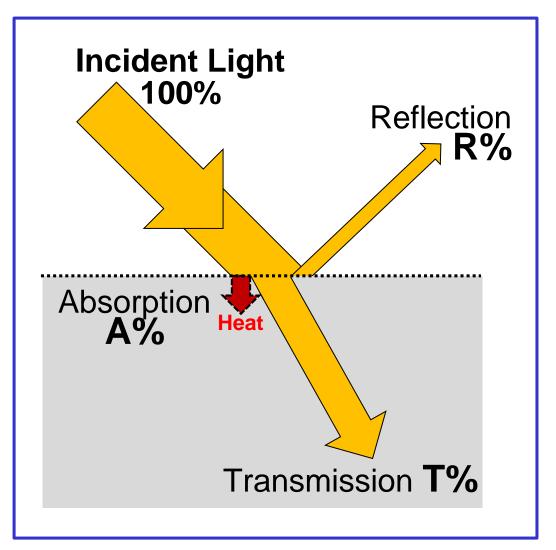






## Light Interaction with Non-Luminescent Matter



<u>Combination</u> of transmission, reflection, and absorption:

## T%+R%+A%=100%

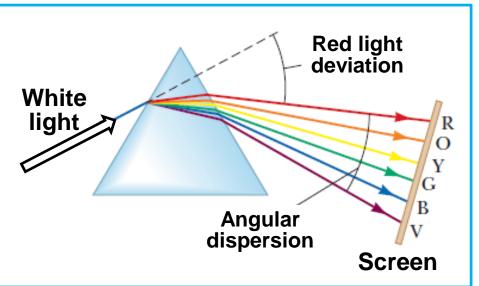
- No material is 100% transparent.
- No material is 100% absorbing either.

## **Dispersion of Light** splitting of light into its component colors

<u>Different colors</u> (wavelengths) of light *travel at different speed in the same material* and therefore <u>refract differently</u>:

- Red (longer wavelength) is bent less.
- Violet (shorter wavelength) is bent more.
- > This allows for <u>separation of colors</u> in certain geometries.

### **Glass prism**

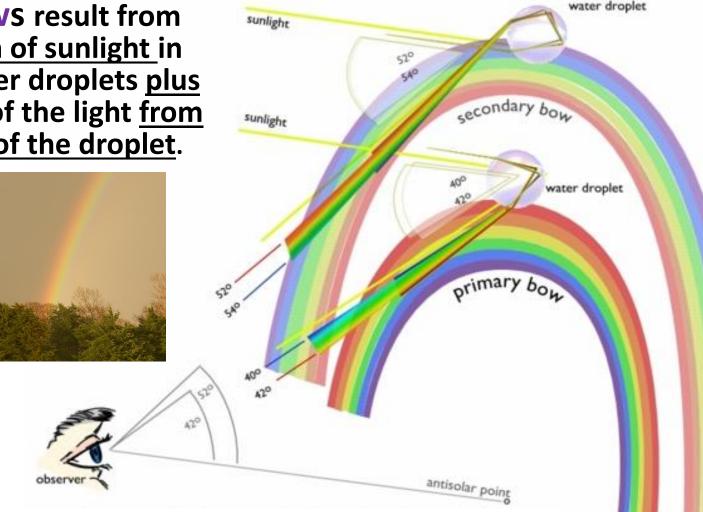


# White Ught raindrop violet indigo blue green yellow orange

Water droplet

# Rainbow

**Rainbows** result from refraction of sunlight in falling water droplets plus reflection of the light from the back of the droplet.



The size of the droplets influences the rainbow appearance: large droplets (>1mm) result in lack of blue color, small droplets make red disappear; fine mist and fog (<0.05mm) produce white or "fog" bow.

# Rainbows...in your backyard!



# All you need is small water droplets and bright sunlight!



#### Can you see the rainbow when the Sun is overhead? Can you see the full circle? Think again 🙂

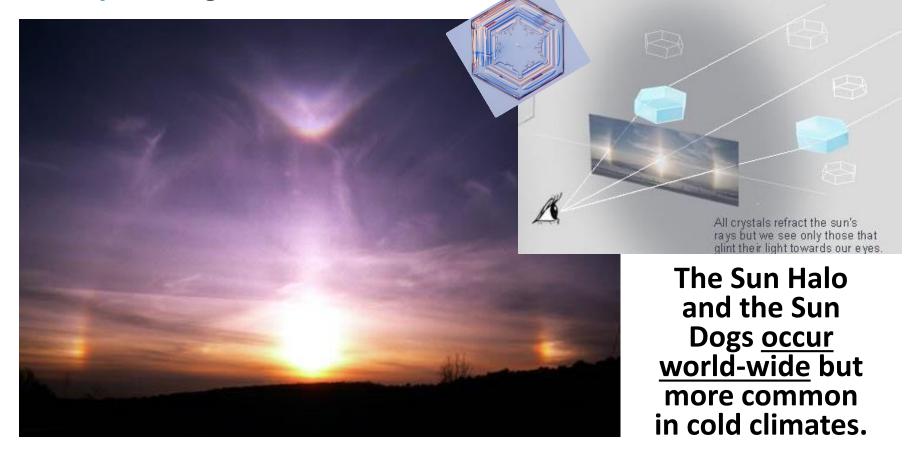
All you need to do İS position yourself between the Sun and the raincloud!

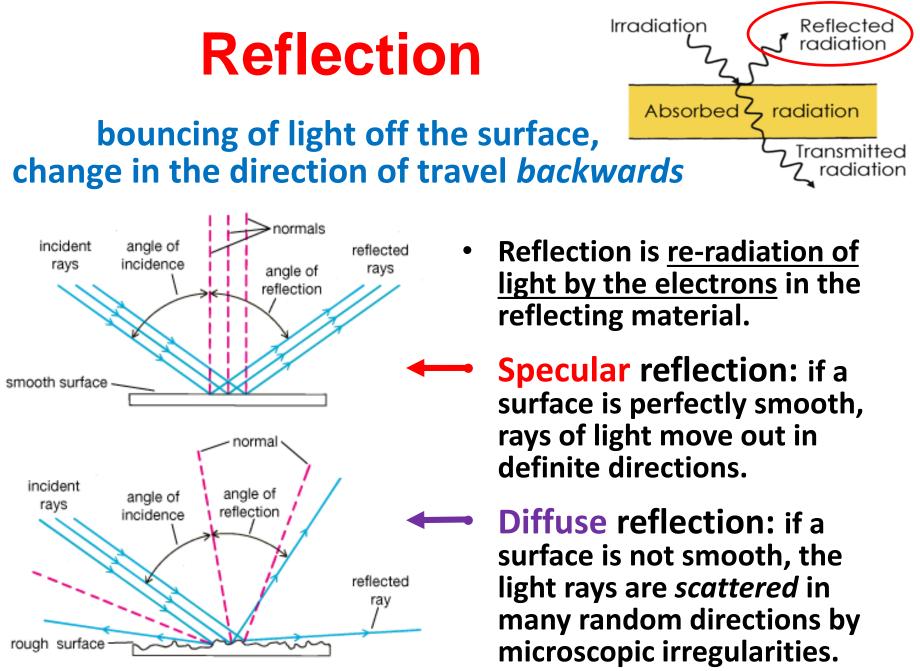


# What happens to light if we have ice crystals in the air instead of water droplets?

## The Sun Halo and the Sun Dogs

## formed by light refraction in horizontally floating hexagonal plate ice crystals high in the cirrus clouds.





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## One-way Mirror (reflection+transmission)



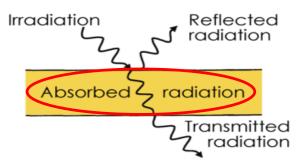
A one-way mirror reflects most of the light that strikes its surface

It also transmits some light to a person behind the mirror in a darkened room.

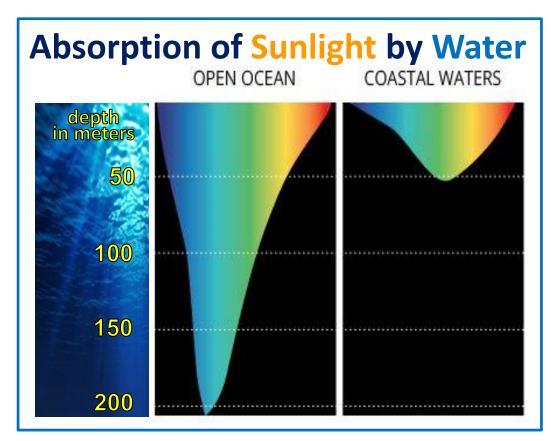


# **Absorption**

### disappearance of a photon

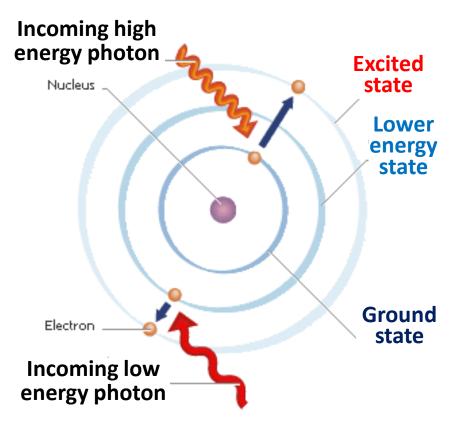


- Absorption of electromagnetic radiation is the process in which the <u>energy of a photon is taken up by matter</u>, typically the electrons of an atom.
- Transparent and translucent objects absorb some part of the incident light.
- Dark opaque objects absorb most of the incident light.
- In most cases, energy of the absorbed photon is converted to *heat*.

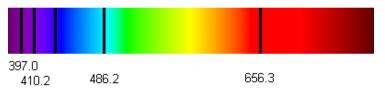


# **Absorption Spectrum**

Absorption of light can happen when the photon energy (i.e. *frequency*) matches one of the allowed transitions between energy levels of that particular atom.

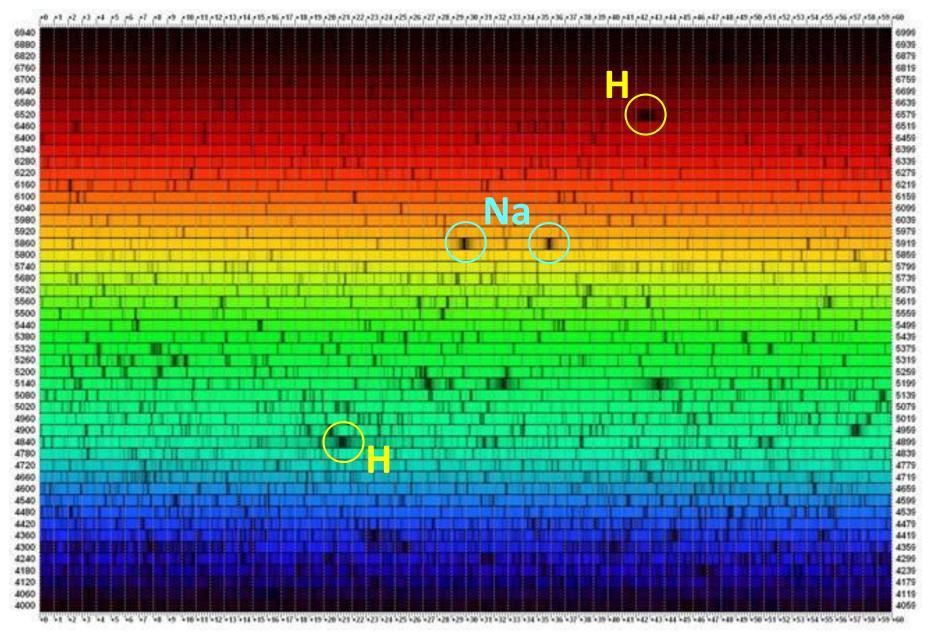


**Example: Hydrogen** 



A star will create an absorption line spectrum because the continuous spectrum emitted by the dense, opaque gas that makes up most of the star passes through the cooler, transparent atmosphere of the star.

## **Absorption Spectrum of the Sun**



## **Sunlight Filtered through Atmosphere**

Absorption of sunlight by various gas molecules that are present in the Earth's atmosphere is seen as absorption bands in the Sun spectrum.

